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## Stereocontrolled Elongation of a Functionalized Isoprene Unit on the E or Z Terminal Methyl of Terpenoids via N-Ylide Rearrangement of the Common Ammonium Salts

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Stereocontrolled elongation of a functionalized isoprene unit on the E or Z terminal methyl of terpenoids was achieved by the N-ylide rearrangement of the common ammonium salts under the selected reaction conditions. One of rearrangement products, all-(E)-sesquiterpene carboxylate was converted into  $\beta$ -sinensal, a component of the essential oil of Chinese orange. General aspects of these transformations were described.

The [2,3] and [3,3] sigmatropic rearrangements are often used for the stereoselective preparation of di- and trisubstituted olefins. Treatment of a quaternary ammonium salt with a base may bring about the formation of an ammonium ylide species followed by spontaneous [2,3] sigmatropic rearrangement to give a homoallylic tertiary amine. [2,3] Sigmatropic rearrangement is occasionally accompanied by Hofmann elimination and other side reactions. Therefore reaction selectivity as well as stereoselectivity must be attained for the synthetic application.

We have reported a [2,3] sigmatropic N-ylide rearrangement that provides (Z)- or (E)-homoallylic dimethylamines with high stereoselectivity in good yields, and [2,3] sigmatropic rearrangement of ylides with a powerful electron-withdrawing substituent (e.g.  $CO_2E$ t or Ac) at the ylide carbon affords exclusively (E)-olefin. Furthermore, the vinylogous ylide derived from quaternary salt 1 affords (Z)-olefin 2, whereas treatment of 1 with an alkoxide base in a protic solvent gives, after hydrolytic work-up, trisubstituted (E)-olefinic aldehyde 4 with complete stereoselectivity  $(Scheme 1)^{4,5}$ 

Br 
$$Me_2N^+$$
  $CO_2Et$   $THF- HMPA$   $Me_2N^+$   $CO_2Et$   $Me_2N^+$   $CO_2Et$   $Me_2N^+$   $CO_2Et$   $Me_2N^+$   $CO_2Et$   $Me_2N^+$   $CO_2Et$   $Me_2N^+$   $CO_2Et$   $CO_2ET$ 

Herein we report a unique [2,3]sigmatropic *N*-ylide rearrangement which provides predominantly (*Z*)- or (*E*)-trisubstituted olefins according to the reaction conditions from the common ammonium salts.

EtOC NMe<sub>2</sub> 
$$\overline{B}$$
r OMe  $\overline{B}$   $\overline{B}$ 

**Table 1.** Reaction of *N*-Tiglyl-*N*- $\beta$ -Methallyldimethylammonium Salt **5** 

Run	Base	Solvent	Conditions(°C/h)	Yield(%)	$Z:E^{\mathbf{b}}$
1	KO <sup>t</sup> Bu	THF-HMPA <sup>C</sup>	-70 / 2	41	27: <b>73</b>
2	KO <sup>t</sup> Bu	THF-DMPU	<sup>1</sup> –70/2	46	28: <b>72</b>
3	KO <sup>t</sup> Bu	THF	<b>-70 / 2</b>	45	30: <b>70</b>
4	KO <sup>t</sup> Bu	DME	<b>-70 / 2</b>	55	39: <b>61</b>
5	KO <sup>t</sup> Bu	CH <sub>2</sub> Cl <sub>2</sub>	$-70 \sim rt / 3$	52	42:58
6	LDA	THF	<b>-70 / 2</b>	45	43: <b>57</b>
7	DBU <sup>e</sup>	THF	–70 ~ rt / 12	66	40: <b>60</b>
8	DBU	EtOH	0/2	60	<b>72</b> :28
9	NaNH <sub>2</sub>	NH <sub>3</sub>	-65 / 4	36	<b>73</b> :27
10	LHMDS <sup>f</sup>	$NH_3$	-65 / 4	46	<b>72</b> :28
	KOEt	EtOH	0/1	68	<b>80</b> :20
12	NaOEt	EtOH	0/1	56	<b>81</b> :19
13	LiOEt	EtOH	0/1	41	<b>87</b> :13

<sup>a</sup> Isolated yield <sup>b</sup> The stereochemistry of the products was confirmed by  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR (CDCl<sub>3</sub>) as follows (only the chemical shifts of olefinic methyl protons for  $^1\mathrm{H}$  NMR and those of olefinic methyl carbons for  $^{13}\mathrm{C}$  NMR are shown): (*E*)-7:  $\delta$  1.61 (3H, s); 15.9, (*Z*)-7:  $\delta$  1.69 (3H, s); 23.8. <sup>c</sup> HMPA content was 20 vol%.  $^d$  1,3-Dimethyl-3,4,5,6-tetrahydro-2(1*H*)-pyrimidinone  $^e$  1,8-Diazabicyclo[5,4,0]undec-7-ene  $^f$  Lithium hexamethyldisilazide

Quaternary ammonium salt 5 was treated with various bases under a variety of reaction conditions. The product mixture was analyzed by GLC and the stereoisomers (E)-7 and (Z)-7 were separated and characterized by  $^{1}$ H- and  $^{13}$ C-NMR spectra. The results are summarized in Table 1.

Treatment of 5 with potassium *tert*-butoxide in a mixture of either THF-HMPA or THF-DMPU gave (*E*)-7 predominantly (runs 1 and 2). This result is in a sharp contrast to that of the rearrangement of salt 1 shown previously. Similarly, the reaction of 5 with a base, especially potassium *tert*-butoxide in an aprotic solvent such as THF, DME or CH<sub>2</sub>Cl<sub>2</sub> afforded (*E*)-7 predominantly (runs 3~5). And higher polarity of solvent seems

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to increase the E-selectivity of the rearrangement.

On the contrary, the stereoselectivity of the [2,3]rearrangement was dramatically reversed and (Z)-7 was formed predominantly when the salt 5 was treated with a base in a protic solvent, without isomerization of the parent E double bond in the tiglate moiety (runs 8~13). Presumably the formed ammonium ylide intermediate 6 rapidly underwent the [2,3] sigmatropic rearrangement without isomerization to a sort of enammonium salt 3 as observed in the case of 1. Furthermore, lower basicity increased the Z-selectivity of the rearrangement (runs 8~13).

This new stereoselective reaction will play an important role in natural product synthesis, especially in the terpenoid field. E-Selectivity of the rearrangement was applied to a stereoselective synthesis of  $\beta$ -sinensal (8), a characteristic component of the essential oil of Chinese orange (Scheme 2).

$$\frac{1) \text{ Ca}(\text{OCl})_2, \text{ CO}_2}{2) \text{ HNMe}_2 \text{ aq.}} \\ \text{NMe}_2 \\ \frac{\text{S-Myrcene}}{\text{CP}_3\text{CN}} \\ 0 \text{ °C, 15 h} \\ \text{Y. quant.} \\ 10 \\ \text{MeO}_2\text{C} \\ \frac{\text{KO}^{1}\text{Bu}}{\text{THF - DMPU}} \\ -70 \text{ °C, 2 h} \\ \text{Y. 69%} \\ \text{MeO}_2\text{C} \\ \frac{\text{NMe}_2}{\text{C}_2\text{Me}} \\ + (Z) \text{-isomer} \\ \frac{\text{E}_2\text{-11}}{\text{C}_2\text{-isomer}} \\ \frac{\text{E}_2\text{-11}}{\text{C}_2\text{-isomer}} \\ \frac{\text{E}_2\text{-11}}{\text{C}_2\text{-isomer}} \\ \frac{\text{E}_2\text{-11}}{\text{C}_2\text{-isomer}} \\ \frac{\text{E}_2\text{-11}}{\text{C}_2\text{-isomer}} \\ \frac{\text{E}_2\text{-11}}{\text{C}_2\text{-isomer}} \\ \frac{\text{E}_2\text{-12}}{\text{C}_2\text{-isomer}} \\ \frac{\text{E}_2\text{-13}}{\text{C}_2\text{-isomer}} \\$$

Scheme 2.

13 R =  $CH_2OH$ 

8 R = CHO

4) MnO<sub>2</sub> (Y.72%)

β-Myrcene was converted via ene-type chlorination<sup>6</sup> followed by amination<sup>7</sup> into internal allylamine 9, which was reacted with methyl γ-bromotiglate<sup>8</sup> in MeCN to give quaternary salt 10. Treatment of 10 with potassium tert-butoxide in a mixture of THF-DMPU resulted in a 71:29 mixture of (E)-11 and (Z)isomer in a 69% combined yield. Notably, isolation of (E)-119 was easily achieved by column chromatography on silica gel.

The dimethylamino group in (E)-11 was removed viaquaternization with MeI, followed by treatment with sodium amalgam in a buffer solution without any conjugate reduction to give  $\alpha,\beta$ -unsaturated ester 129, which was reduced to the corresponding unsaturated alcohol 139 in 90% yield with AlH<sub>3</sub>10 prepared in situ from aluminum chloride and lithium aluminum hydride.

Treatment of 13 with active manganese (IV) oxide in hexane at 0 °C for 3 h furnished the desired  $\beta$ -sinensal (8) in 72% yield.

As summary, stereocontrolled elongation of a functionalized E isoprene unit on the E or Z terminal methyl of terpenoids was achieved by the N-ylide rearrangement of the common ammonium salts under the selected reaction conditions. The present method is useful for the construction of both (E, E) and (E, Z) terpenoids.

## References and Notes

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- Spectral data for the selected compounds are as follows. (E)-11:  ${}^{1}\text{H-NMR}$  (CDCl<sub>3</sub>)  $\delta$  1.60(s, 3H), 1.82(s, 3H), 2.0-2.4(m, 6H), 2.29(s, 6H), 3.3-3.4(m, 1H), 3.73(s, 3H), 4.9-5.0(m, 2H), 5.04(d, J = 10.9 Hz, 1H), 5.1-5.3(m, 1H),5.20(d, J = 17.8 Hz, 1H), 6.36(dd, J = 10.9, 17.8 Hz, 1H),and 6.67(d, J = 10.2 Hz, 1H). 12: <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ 1.62(s, 3H), 1.84(s, 3H), 2.0-2.7(m, 8H), 3.73(s 3H), 5.01(m, 2H), 5.06(d, J = 10.9 Hz, 1H), 5.1-5.3(m, 1H), 5.24(d, J = 17.8 Hz, 1H), 6.38(dd, J = 10.9, 17.8 Hz, 1H),and 6.75(t, J = 10 Hz, 1H). 13: IR(neat) 3350, 1040, 990, and 900 cm<sup>-1</sup>;  ${}^{1}H$ -NMR (CDCl<sub>3</sub>)  $\delta$  1.61(s, 3H), 1.67(s, 3H), 2.0-2.7(m, 8H), 3.99(s 2H), 5.01(m, 2H), 5.06(d, J =10.9 Hz, 1H), 5.1-5.3(m, 1H), 5.24(d, J = 17.8 Hz, 1H), 5.40(t, J = 8 Hz, 1H), and 6.38(dd, J = 10.9, 17.8 Hz, 1H).
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